



Mineral Properties

I. Nonferromagnesian Silicates

Quartz

SiO_2

Silicon Dioxide

Hardness 7.5

Hexagonal crystals

No cleavage

Conchoidal fracture

Variable color varieties

Uses: electronics, lenses, jewelry, main component of sand

Plagioclase

Sodium feldspar

H = 6

Perfect cleavage in 2 directions not quite at 90 deg.

irregular fracture on ends

variable color

Uses: ceramics, common Simatic rock forming mineral

Weathers to form clay

Shows striations on one crystal face in good samples

Orthoclase

$\text{K}(\text{AlSi}_3\text{O})$

Potassium feldspar

H = 6

Perfect cleavage in 2 directions near 90°

Irregular fracture on ends of crystals

Uses: ceramics, coating for paper, common Sialic rock former

Weathers to form Kaolinite clay minerals

Muscovite (white mica) (isenglas)

$\text{KAl}_3\text{Si}_3\text{O}_{10}(\text{OH})_2$ Structural family: Phyllosilicate

H = 2-2.5

Cleavage good in one direction, flexible

Uses: electronic dielectric, heat resistant gaskets, paint,

used as windows before plateglass

Mined extensively near Spruce Pine and Kings Mountain

II. Ferromagnesian Silicates

Hornblende

Complex ferromagnesian silicate

Structural family: amphibole (double chain silicate tetrahedra)

H = 5 - 6

Elongated crystals

C - Black ->green, S - pale green

Common in granitic, basaltic and metamorphic rocks

Augite

Complex ferromagnesian silicate

Structural family: pyroxene (single chain silicate tetrahedra)

H - 5.6

Stubby crystals

C - blackish green to pale green, S - pale green

Found in basalt, peridotite and high grade metamorphic rocks

Olivine

$(\text{Mg,Fe})_2\text{SiO}_4$ independent silicate tetrahedra bound by metals

H = 6.5 - 7

uneven fracture or granular masses

C - green, S - white

Common in peridotite and basalt

Gemstone (peridot), refractory brick

Biotite (black mica)

Complex ferromagnesian silicate

Cleavage perfect in one plane

H = 2.2 - 2.5

C - black, brown, green; S - white to gray

mineral of granite and other igneous rocks, metamorphic rocks

III. Other Silicates

Garnet

Complex aluminum silicates

No cleavage

H = 5.6 - 7.5

C - deep red (almandine), white, yellow, brown (grossular)

S - White

Found in metamorphic rocks

Gemstones (January), abrasives

Chlorite

Complex Iron, magnesium, silicon, aluminum, hydroxide

Perfect fine scale cleavage

H = 2.0 - 2.5

C - green, S - Gray, white, pale green

Common low grade metamorphic accessory mineral

Talc (soapstone)

Hydrous Magnesium Silicate

H = 1

Formed from alteration of augite, olivine or hornblende

Used in talcum powder for cosmetics and the rubber industry

IV. Sulfides

Pyrite (fool's gold)

FeS₂ Iron Sulfide

Cubic crystals, uneven fracture, striated faces

H - 6 - 6.5

C - pale brass yellow, paler than Chalcopyrite

S - greenish black

metallic luster, unlike gold pyrite is brittle

The most common sulfide mineral, Ig, Met, Sed, Hydrothermal veins

Ore of Sulfur

Chalcopyrite

CuFeS₂ Copper, Iron Sulfide

Uneven fracture

H = 3.5 - 4.5

C - Brass yellow, S - greenish black

metallic luster, softer than pyrite

Copper ore, forms in hydrothermal veins

Sphalerite

ZnS (Zinc Sulfide)

Perfect cleavage in 6 directions at 120°

H = 3.5

C - Shades of brownish to reddish resinous metallic luster

S - Reddish brown and smells of sulfur when streaked vigorously

Occurs with galena and pyrite in hydrothermal veins

The principle ore of Zinc

Galena

PbS (Lead Sulfide)

Perfect cleavage in three directions all at 90° (cubic)

H = 2.5

C = metallic gray or silver, S - Gray

Specific gravity 7.6 (heaviest of our mineral samples)

Most important lead ore, contains silver

Occurs in hydrothermal veins

V. Sulfates

Gypsum (Alabaster, Selenite, Satin Spar)

CaSO₄ · H₂O (Hydrated Calcium Sulfate)

Tabular Crystals, fibrous, Massive or granular

H - 1 - 2.5

C - White, pearly; S - white

Sheets (Selenite), Fibrous (Satin Spar), Massive (Alabaster)

Sedimentary EVAPORITE deposit

VI. Oxides

Corundum

Al_2O_3 (Aluminum oxide)

Short hexagonal crystals

H - 9 (hardest except for diamonds)

C - Light gray, blue, red etc. S - none (cuts streak plate)

Gems (rubies and sapphires), Abrasives, Laser optics

Occurs in metamorphic and igneous rocks

Hematite

Fe_2O_3 (Iron Oxide)

Specular - Metallic gray, granular or massive

Oolitic - Rusty red masses, beady, earthy

H = 2.5

Common in all types of rocks. Causes red color in soils and rock

An ore of Iron

Magnetite

Fe_3O_4

Uneven fracture, granular masses

H = 5.5

C - Iron black, S - Iron black

Metallic luster, strongly magnetic

An ore of Iron, found in most igneous rocks

VII. Carbonates

Calcite

CaCO_3 (Calcium Carbonate)

Perfect rhomboid cleavage

H = 3

C - usually clear to white, some tinting, S - White

Effervesces strongly in HCl

Main component of limestone, dolomite and marble

In cave deposits and travertine

Dolomite

$\text{CaMg}(\text{CO}_3)_2$ (Calcium Magnesium Carbonate)

Cleaves into rhombs or occurs in granular masses

H = 3.5 - 4

C - white, pink, gray, brown; S - white to pale gray

Effervesces if pulverized or in hot dilute HCl
Makes up dolomitic rock

VIII. Native element minerals

Graphite

C (pure carbon)
Foliated scaly or earthy masses
H = 1 - 2
C = Steel gray to black, S = gray or black
Occurs in metamorphic rocks
Feels greasy, marks on paper
Used as a dry lubricant, electrical applications, sporting equipment, high temperature crucibles

Sulfur

S
Bright yellow masses
Smells of sulfur

IX. Halides

Halite (rock salt)

NaCl (Sodium Chloride)
Perfect cubic cleavage or granular masses
H = 2.5 - 3
C = white, clear, pale colors; S - white
Salty taste, soluble in water
A sedimentary EVAPORITE mineral
Many uses

Fluorite

CaF₂ (Calcium Fluoride)
Octahedral and cubic crystals
H = 4
C - white, yellow, green, purple; S - white
Cleaves easily, vitreous, transparent to translucent
Would be nice gems but too soft
Source of Fluorine for toothpaste and mouth rinse, hydrofluoric acid, a flux for steel making
Fluoresces in UV light

X. Mineraloids

Kaolinite

Hydrated aluminum silicate, a clay mineral
Derived from the weathering of feldspar
Mined in North Carolina for the ceramics industry

Bauxite

Aluminum Oxide
C - Reddish to brown with small round concretions in it
Component of Lateritic soils of the tropical rainforests
The principle ore of the Aluminum industry

Limonite

Iron Oxide (rust)
Rusty yellowish brown
Redeposited iron oxide derived from weathering of hematite
Ore of Iron

XI. Phosphates

Apatite
Calcium fluorophosphate
H = 5
Source of fertilizer due to calcium and phosphate content
Some gem use but too soft for much value